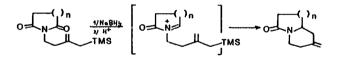
CYCLISATION OF ALLYL SILANES. FORMAL TOTAL SYNTHESIS OF (+)-MESEMBRINE

Jean-Claude Gramain and Roland Remuson

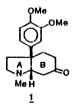
Laboratoire de Chimie et Biochimie des Substances Naturelles, Unité Associée au CNRS n° 485, Université de Clermont-II, B.P. 45, 63170 Aubière, France.

Abstract : The intramolecular reaction of an α -acyl iminium ion with an allyl silane occurs under non acidic conditions to afford the key intermediate 8 for the synthesis of (\pm) -mesembrine 1 in a stereospecific manner.

We recently reported a new approach to the synthesis of indolizidine and quinolizidine bicyclic systems¹ based upon the electrophilic reaction of α -acyl iminium ions with allyl silanes. An important feature of these cyclisations was the formation of an exocyclic double bond allowing further functionalisation of the heterocyclic ring (Scheme 1).



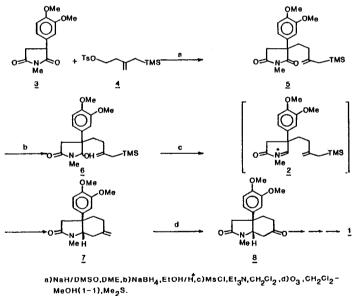
Scheme 1



We wish, in the present communication, to describe the application of the above methodology towards an efficient stereospecific synthesis of (\pm) -mesembrine <u>1</u>. The preparation of this alkaloid and its congeners is of considerable interest² due to the CNS activity of this class of compounds and to their structural similarity to the *Sceletium*, *Amaryllidaceae*³ and even morphine type alkaloïds⁴.

The key step in our strategy involves the intramolecular electrophilic cyclisation of the *in situ* generated α -acyl iminium intermediate 2 to give the B ring of 1. Examination of the Dreiding models showed that the geometry of side chain relative to the flat five membered ring of 2 should greatly favour the formation of *cis* fused ring as required in mesembrine 1.

Alkylation of the anion of <u>3</u> (NaH/DMSO ; DME, 50°C, 1 h) with the tosylate $\underline{4}^6$ led to the imide <u>5</u> (60 %)^{7,8} exclusively (Scheme 2). The reduction of <u>5</u> using Speckamp's method⁹ (NaBH₄, EtOH, HCl) proceeded with high regioselectivity affording the 2-hydroxy lactam <u>6</u> (60 %)¹⁰ as a mixture of diastereomers¹¹.



Scheme2

It was then expected that the treatment of <u>6</u> with CF_3CO_2H/CH_2CI_2 would generate <u>2</u> which would in turn react with the allyl silane component to produce the olefin <u>7</u>. However, under these conditions cyclisation was followed by the isomerisation of the exocyclic double bond to the more stable endocyclic position.

Non acidic conditions¹² were thus employed in order to avoid this undesirable isomerization process. It was found that, by treatment of <u>6</u> with MsCl/Et₃N in CH₂Cl₂ at room temperature overnight, the desired bicyclic olefin $\underline{7}^{13}$ was obtained in a stereospecific manner in 80 % yield (with respect to consumed starting material, 60 % conversion).

Ozonolysis¹⁴ of $\underline{7}$ (O₃, CH₂Cl₂-MeOH (1-1), - 78°C, Me₂S) led to the keto lactam $\underline{8}^{15}$ in nearly quantitative yield. The spectral data for this compound were identical in all respects with previously reported literature data^{2e}. The expected *cis* stereochemistry of compound $\underline{7}$ is thus confirmed.

In conclusion, the synthesis of $\underline{8}$ was realized in four steps (30 % overall yield) from the readily avalaible imide $\underline{3}$. Since the conversion of the intermediate $\underline{8}$ to $(\underline{+})-\underline{1}$ has been reported to proceed in nearly quantitative yield, this approach constitutes a new formal synthesis of this alkaloid.

Acknowledgements

High resolution mass spectra were recorded in the Mass Spectroscopy Laboratory, University of P. and M. Curie, Paris. We thank Pr. D. Besserre (NMR Service of CRMP) for NMR spectra. The authors would also like to thank B. Perrin for technical assistance.

References and notes

- 1. J.C. Gramain and R. Remuson, Tetrahedron Lett., 1985, 26, 327.
- ^aM. Shamma and H.R. Rodriguez, Tetrahedron Lett., 1965, 4847; ^bR.V. Stevens and M.P. Wentland, J. Am. Chem. Soc., 1968, <u>90</u>, 5580; ^cS.L. Keely, Jr and F.C. Tahk, *Ibid*, 1968, <u>20</u>, 5584; ^dT.J. Curphey and H.L. Kim, Tetrahedron Lett., 1968, 1441; ^eH. Taguchi, T. Oh-Ishi and H. Kugita, Chem. Pharm. Bull., 1970, <u>18</u>, 299; ^fT. Oh-Ishi and H. Kugita, *Ibid*, 1970, <u>18</u>, 1008; ^gG. Otani and S. Yamada, *Ibid*, 1973, <u>21</u>, 2130; ^hM. tanglois, C. Guillonneau, J. Heingan and J. Maillard, Tetrahedron, 1971, <u>27</u>, 5641; ⁱJ.B.P.A. Wijnberg and W.N. Speckamp Tetrahedron Lett., 1975, 3963; ^jJ.B.P.A. Wijnberg and W.N. Speckamp, Tetrahedron, 1978, <u>34</u>, 2579; ^kP.W. Jeffs, R. Redfearn and J. Wolfram, J. Org. Chem., 1983, <u>48</u>, 3861; ¹K.S. Kochlar and H.W. Pinnick, Tetrahedron Lett., 1983, <u>24</u>, 4785; ^mI.H. Sanchez, M.I. Larraza, H.J. Flores, I. Rojas and R. Alcala, Synthetic Communications, 1983, 13, 35.
- R.V. Stevens in "The Total Synthesis of Natural Products", Vol. III, Ed. J. ApSimon, J. Wiley and Sons, New-York, 1977, p. 453.
- D. Lednicer and L.A. Mitscher in "The Organic Chemistry of Drug Synthesis", Vol. II, J. Wiley and Sons, New-York, Ch. 10, p. 314.
- 5. C.A. Miller and L.M. Long, J. Am. Chem. Soc., 1951, 73, 4895.
- Tosylate <u>4</u> was prepared from the corresponding alcohol¹ using standard conditions (TsCl, pyridine, overnight).

- 7. All NMR spectra were recorded at 60 MHz (1 H) and 15 MHz (13 C).
- 8. Imide $\underline{5}$: oil, IR (film): 1780, 1710 cm⁻¹; ¹H NMR (CDCl₃): δ 6.9 (m, 3H); 4.6 (s, 2H); 3.85 (s, 3H); 3.9 (s, 3H); 3.05 (s, 5H); 2.1 (m, 4H); 1.55 (s, 2H); 0.0 (s, 9H); ¹³C NMR (CDCl₃): δ 181.7; 176.9; 150.5; 149.8; 147.55; 133.9; 119.8; 112.5; 111.2; 108.9; 57.3; 52.6; 42.7; 39.8; 34.4; 28.3; 26.3; 0.0; Calc. for C₂₁H₃₁NO₄ Si 389.2014; found 389.2025.
- 9. J.C. Hubert, J.B.P.A. Wijnberg and W.N. Speckamp, Tetrahedron, 1975, 31, 1437.
- 10. Hydroxylactam <u>6</u>: oil, IR (film): 3350, 1710, 1685 cm⁻¹; ¹H NMR (CDCl₃): δ 6.9 (m, 3H); 6.2 (m, 1H); 5.2 (m, 1H); 4.6 (m, 2H); 3.9 (s, 6H); 2.9 (s, 5H); 1.3-2.1 (m, 4H); 1.5 (s, 2H); 0.0 (s, 9H).
- 11. The espected regiomer <u>6</u> (60 %) was obtained as a mixture of the two possible stereomers in a 9:1 ratio. They were identified by the multiplicity of the CH-OH signal and the presence of two N-CH₃ singlets at 2.9 and 3.0 ppm in the ¹H NMR spectra. Both diastereomers were isolated by column chromatography and the spectral data of the major diastereomer is given in note 10. It is in agreement with these mentioned by J.B.P.A. Wijnberg and W.N. Speckamp (Tetrahedron, 1978, <u>34</u>, 2579) for analogous products. The other regiomer (corresponding to the reduction in 5 position of the imide group) was also

isolated (20 %) as a mixture of diastereomers and identified by the multiplicity (broad multiplet) of the CHOH signal in 1 H NMR.

- 12. a A.R. Chamberlin and J.Y.L. Chung, Tetrahedron Lett., 1982, <u>23</u>, 2619 ; b A.R. Chamberlin and J.Y.L. Chung, J. Am. Chem. Soc., 1982, <u>105</u>, 3653 ; c A.R. Chamberlin, H.D. Nguyen and J.Y.L. Chung, J. Org. Chem., 1984, 49, 1682.
- 13. Compound $\underline{7}$: oil, IR (film) 1690 cm⁻¹; ¹H NMR (CDCl₃): δ 6.9 (m, 3H); 4.85 (s, 2H); 4.05 (t, 1H); 3.9 (s, 6H); 2.9 (s, 3H); 2.6 (m, 4H); 2.0 (m, 4H); ¹³C NMR (CDCl₃): δ 173.7; 118.4; 111.3; 110.7; 110.2; 64.4; 56.1; 56.0; 47.2; 35.0; 33.5; 29.7; 26.9; Calc. for C₁₈H₂₃NO₃: 301.1672; found: 301.1676.
- 14. M.G. Silvestri, J. Org. Chem., 1983, 48, 2419.
- 15. Compound <u>8</u>: oil, IR (CHCl₃) 1720, 1685 cm⁻¹; ¹H NMR (CDCl₃): 6 6.9 (m, 3H); 4.4 (t, 1H); 3.9 (s, 6H); 2.9 (m, 7H); 2.35 (m, 4H). (Received in France 18 April 1985)